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PHY 248 Lecture 16

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3/5/07

$$k = 8.98755 \times 10^9 \text{ Nm}^2/\text{C}^2$$

$$V = -\frac{k e^2}{r}$$

just like
gravity $\overset{\circ}{p}$ We'll solve this problem 4
in Phy 448 (Junior year)

Energy

$$E = \frac{p^2}{2m} + V(r)$$

$$= \frac{p^2}{2m} - \frac{k e^2}{r}$$

Estimate the ground state energy using the uncertainty principle.

$$\overline{p^2} \sim (\Delta p)^2 \geq \frac{\hbar^2}{(\Delta x)^2} \sim \frac{\hbar^2}{r^2}$$

$$\Rightarrow E = \frac{\hbar^2}{2mr^2} - \frac{k e^2}{r}$$

$$\text{Minimize } E \Rightarrow \frac{dE}{dr} = 0 \Rightarrow -\frac{\hbar^2}{mr^3} + \frac{ke^2}{r^2} = 0$$

$$\Rightarrow r_m = \frac{\hbar^2}{ke^2 m}$$

$$E_m = -\frac{k^2 e^4 m}{2\hbar^2} = -13.6 \text{ eV}$$

Exactly the right answer \Rightarrow dimensional analysis

Not the most convenient units for atomic physics...

Define

$$\alpha = \frac{ke^2}{\hbar c} \approx \frac{1}{137}$$

small
 \Rightarrow perturbation theory

$$a_0 = r_{\min} = \frac{\hbar}{mc} \frac{hc}{ke^2}$$

$$= \frac{\hbar}{mc} \frac{1}{\alpha}$$

Bohr radius

$$E_n = -\frac{m k^2 e^4}{2 \hbar^2 n^2} = -\frac{mc^2}{2} \alpha^2 \frac{1}{n^2}$$

Wave function (Ground state) of e^-

$$\Psi = C e^{-zr/a_0}$$

Although $\Psi = \max$ at $r = 0$, probability per unit r is

$$4\pi r^2 |\Psi|^2 dr = P(r) dr$$

Can show that max of $P(r)$ is at $r = a_0/2$

$$\frac{d}{dr} P(r) = 0 \Rightarrow r^2 \left(-\frac{2z}{a_0}\right) + 2r = 0$$

$$\Rightarrow r = a_0/2$$

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Keeping track of the factors of Z

$$E = \frac{p^2}{2m} + V(r) = \frac{p^2}{2m} - \frac{Zke^2}{r}$$

Uncertainty principle

$$\Rightarrow E = \frac{\hbar^2}{2mr^2} - \frac{Zke^2}{r}$$

$$\frac{dE}{dr} = 0 \Rightarrow -\frac{\hbar^2}{mr^3} + \frac{Zke^2}{r^2} = 0$$

$$\Rightarrow r_m = \frac{\hbar^2}{Zke^2 m} = \frac{a_0}{Z}$$

$$E_m = \frac{\frac{\hbar^2}{2m} Z^2}{a_0^3} - \frac{Zke^2 Z}{a_0}$$

$$= \frac{Z^2}{a_0^2} \left[\frac{\frac{\hbar^2}{2m}}{Z^2} - ke^2 a_0 \right]$$

$$E_n = -\frac{mc^2}{2} \frac{Z^2 \alpha^2}{n^2}$$

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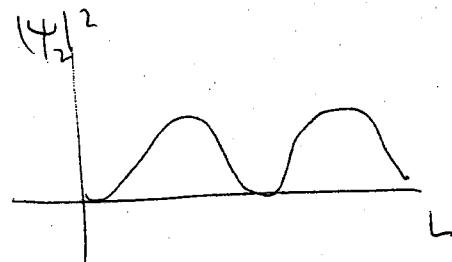
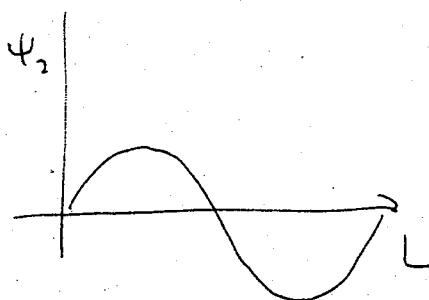
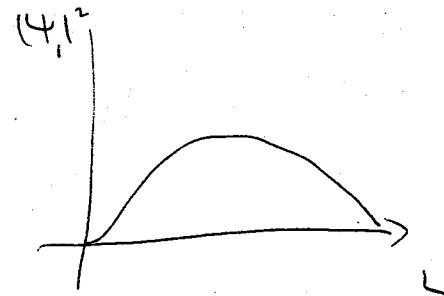
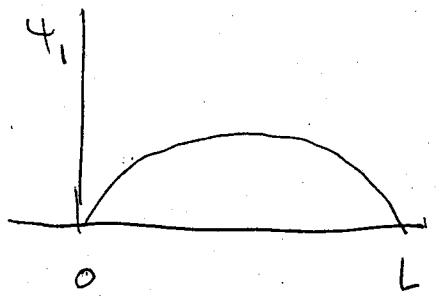
Transition from Quantum to Classical
occur when n (quantum number) becomes large

For example

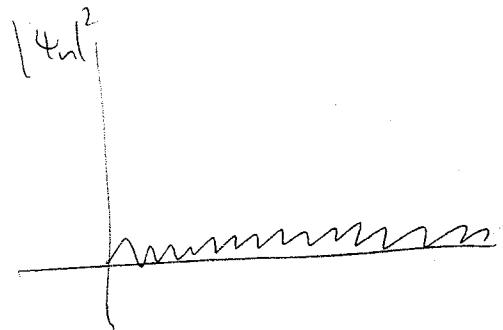
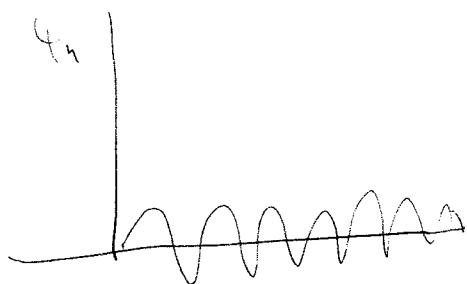
Infinite well

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L} \quad n=1, 2, 3, \dots$$

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2m L^2}$$



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As $n \rightarrow \infty$ 

$|\Psi_n|^2 \approx \text{constant}$ (couldn't tell the probability is changing)

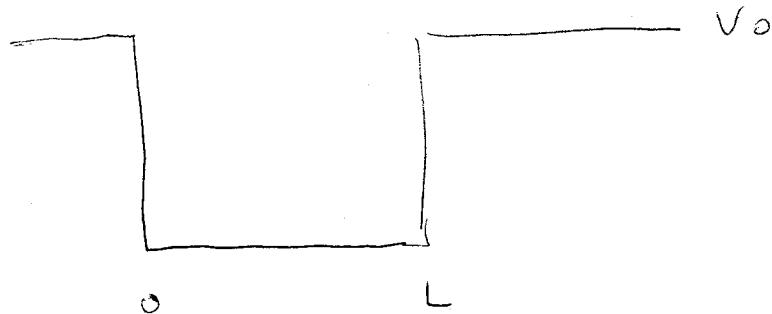
$$|\Psi_n|^2 = \frac{2}{L} \sin^2 \frac{n\pi x}{L} \xrightarrow{\text{time average}} \frac{2}{L} \times \frac{1}{2} = \frac{1}{L}$$

Does it make sense?

$$P(x)dx \propto \frac{dx}{\sqrt{\dots}} \approx \text{constant}$$

$$\text{Properly normalized} \Rightarrow P(x) = \frac{1}{L}$$

Finite Well



We argued that $\Psi(0) = \Psi(L) = 0$ because

$V = \infty$ at $x=0, L$

For $V_0 = \text{finite}$, the wavefunction are not vanishing at $x=0, L$

$$\Psi''(x) = \frac{2m}{\hbar^2} (V - E) \Psi(x)$$

$$\text{For } 0 < x < L \quad V = 0 \Rightarrow \Psi'' = -\frac{2mE}{\hbar^2} \Psi(x)$$

↑ negative

$$\Rightarrow \Psi = A \sin kx + B \cos kx$$

or $A' e^{ikx} + B' e^{-ikx}$

For x outside the well

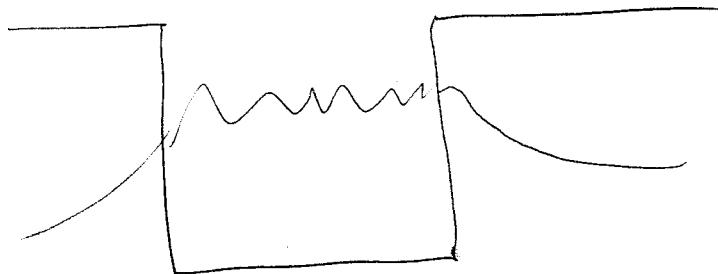
$$\Psi'' = \underbrace{\frac{2m}{\hbar^2} (V_0 - E)}_{\text{positive}} \Psi(x)$$

$$\propto^2 = \frac{2m}{\hbar^2} (V_0 - E)$$

$$\Psi(x) = C e^{kx} + D e^{-kx}$$

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Hence qualitatively



How could this be possible?

Outside the well "kinetic energy" is -ve

Answer : Uncertainty principle

$$\Psi(x) \sim e^{-\alpha x} \Rightarrow \Delta x \sim \frac{1}{\alpha}$$

$$\Rightarrow \Delta p > \frac{\hbar}{\Delta x} \sim \hbar \alpha$$

Large uncertainty in kinetic energy

$$\frac{(\Delta p)^2}{2m} = \frac{\hbar^2 \alpha^2}{2m} = V_0 - E \quad \leftarrow \begin{array}{l} \text{prevent us} \\ \text{from measuring} \\ \text{-ve kinetic} \\ \text{energy} \end{array}$$