Welcome to Physics 202

Today's Topics

The Physics 202 Team

Course Formality and Overview

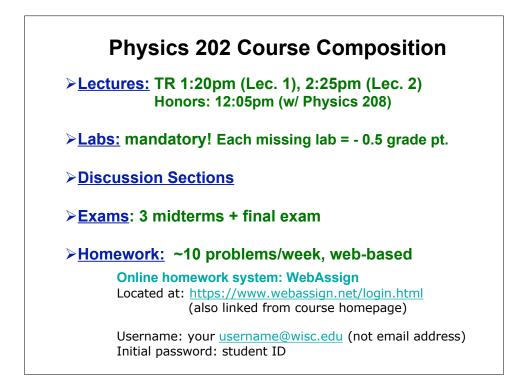
Ch. 23-I: Electric Charge, Coulomb's Law

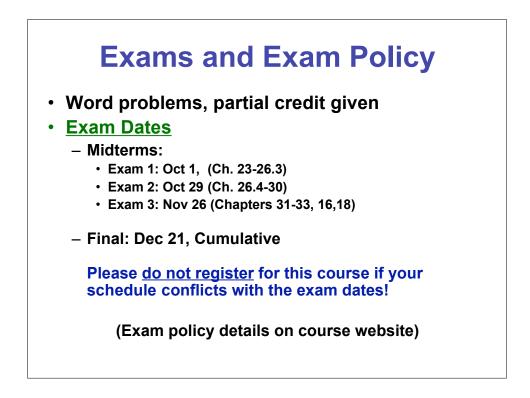
Text: Serway and Jewett, Physics for Scientists and Engineers, 6th ed.

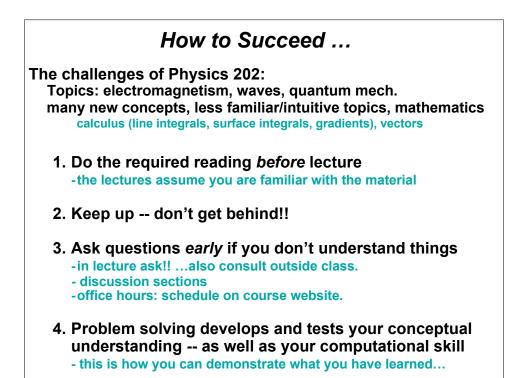
Physics 202 Homepage

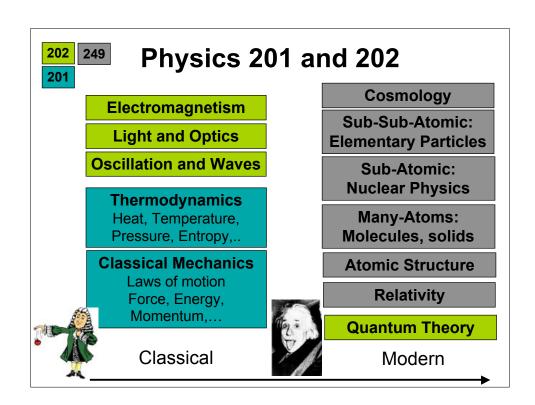
http://www.icecube.wisc.edu/~karle/courses/phys202/ (linked from http://www.physics.wisc.edu/undergrads/undergrad.html)

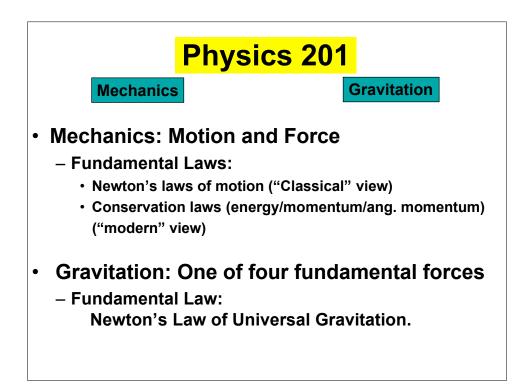
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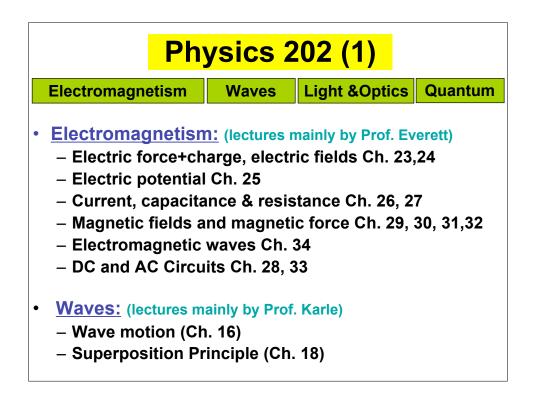


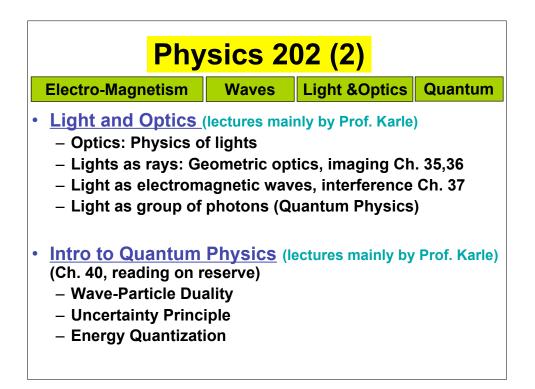












Chapter 23: Electric Fields

Today:

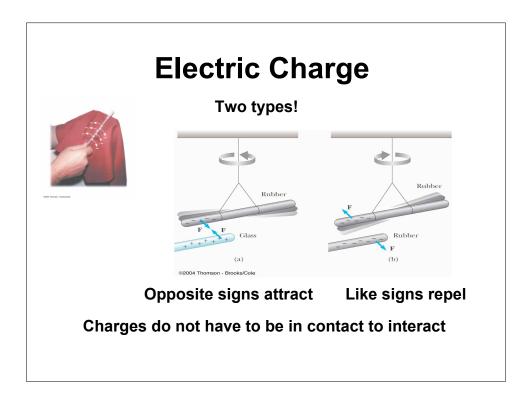
Electric charges

Electric force and Coulomb's Law

Thursday:

- More on Coulomb's Law
- Electric Field
- Exercises

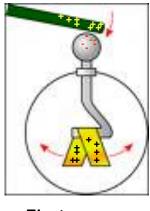
Please read Ch. 23 before Thursday's lecture



Properties of Electric Charge 2(+1) types: positive, negative (+ neutral*). Unit: Coulomb (C). 1 C= chg of 6.24x10¹⁸ protons Building blocks of matter: Mass (kg) Charge (C) -e=-1.602x10-19 9.11x10⁻³¹ Electron +e=+1.602x10-19 1.673x10⁻²⁷ Proton 1.675x10⁻²⁷ Neutron 0 Electric charge is <u>quantized</u>: q=±Ne (e=1.602x10⁻¹⁹ C) Electric charge of isolated system is <u>conserved</u> *Neutral: no charge or equal amount of + and -

Conductors v. Insulators

Consider how charge is carried on macroscopic objects. In Physics 202, we are concerned with only 2 types:



Electroscope

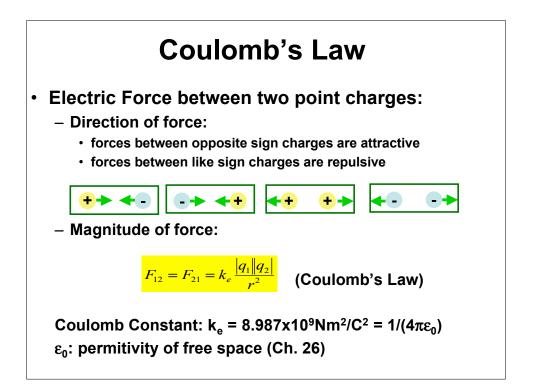
Insulators (glass, plastic, rubber...):

charges NOT free to move

<u>Conductors</u> (metals...):

charges free to move

charge also by induction! (phenomenon: polarization)



Gravitational v. Electric Force

• Proton and electron in a hydrogen atom: $q_e = -1.6 \times 10^{-19} C$ $q_p = 1.6 \times 10^{-19} C$ $r = 5.3 \times 10^{-11} m$ \rightarrow $F_E = 8.2 \times 10^{-8} N$

> The electric force is huge!

- Compared to mass of proton: 1.673x10⁻²⁷ kg
- Compared to gravitational force b/w proton+electron:

→
$$F_G = 3.6 \times 10^{-47} N$$
 (recall: $F_G = \frac{Gm_1m_2}{r^2}$)

Four fundamental forces:
 Strong > Electromagnetic > Weak >> Gravitation

Coulomb's Law: Vector Form

